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containing unknown amounts of  $\text{BaO(s)}$  and  $\text{CaO(s)}$  and placed the sample in a 1.50-L flask containing  $\text{CO}_2(\text{g})$  at  $30.08^\circ\text{C}$  and 750. torr. After the reaction to form  $\text{BaCO}_3(\text{s})$  and  $\text{CaCO}_3(\text{s})$  was completed, the pressure of  $\text{CO}_2(\text{g})$  remaining was 230. torr. Calculate the mass percentages of

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